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Journal of Energy Chemistry



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Highlight Advanced metal sulfide anode for potassium ion batteries

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ARTICLE INFO

Article history: Received 10 December 2017 Revised 11 December 2017 Accepted 12 December 2017 Available online 14 December 2017

Keywords: Potassium ion batteries Metal sulfide Graphene Energy materials

ABSTRACT

Potassium-ion batteries (KIBs) are a promising alternative to lithium-ion batteries owning to the abundance of potassium on Earth and the relatively low K/K^+ redox couple. To date, KIBs remains its infancy and the investigation of anode materials mainly focused on carbon-based materials, which deliver limited reversible capacity. Hence, it is imperative to explore alternative anode materials with high reversible capacity for KIBs. Recently, a pioneering work from Chen's group reported a nanocomposite of Sb₂S₃ nanoparticles anchored on porous S,N-codoped graphene (denoted as Sb₂S₃-SNG) as an advanced anode material for KIBs, which exhibited remarkable enhancements of both capacity and cycling stability, highlighting the rational structure design of Sb₂S₃-SNG for maximum utilization of Sb₂S₃ nanoparticles and graphene layers for energy storage applications in high-performance KIBs.

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With potassium being both abundant and relatively low redox potential (-2.93 V vs. NHE), potassium ion batteries (KIBs) is attracting significant attention with a viewpoint to material suitability and commercial availability [1,2]. Although carbon-based materials have been the dominant anodes in KIBs for their low cost and excellent safety, the limited reversible capacity and rapid capacity fading imped their wide application in KIBs, which drives to explore emerging anode materials with high reversible capacity for energy-dense KIBs [1]. Metal sulfides represent promising anode materials for lithium/sodium-ion batteries because of superior theoretical capacity and redox reversibility to their metal oxide counterparts [3,4]. When the metal sulfides are employed in KIBs, their large volume variations and severe particle aggregation associated with the conversion reactions during cycling induce electrode pulverization and loss of interparticle contact, which consequently result in poor cycling stability [5].

To address the above-mentioned issues, integrating metal sulfide nanoparticles with graphene is recently proposed by Chen and co-workers with a published paper in the Science China Chemistry [6]. This pioneer work described a facile hydrothermal method to obtained advanced nanocomposite of Sb₂S₃ nanoparticles anchored on porous S, N-codoped graphene (denoted as Sb₂S₃-SNG) with attempts to simultaneously address volume fluctuation and reaction kinetics of Sb₂S₃ (Fig. 1a). The surface properties of graphene cles during repeated cycling. Moreover, the anchored nanoparticles act as pillars to reduce the degree of restacking of graphene sheets. This all-in-one approach enables the Sb₂S₃-SNG composite with remarkable enhancements of both capacity and cycling stability when compared with other reported anodes in KIBs (Fig. 1c). An initial charge capacity of 537 mAh g⁻¹ with a capacity retention of 89.4% after 100 cycles was obtained at a current density of 50 mA g⁻¹ in K half-cells. Moreover, the Sb₂S₃-SNG/KVPO₄F-C full cell delivered an initial capacity of 532 mAh g⁻¹ with an average output voltage of ~2.5 V (Fig. 1d), which rendered an energy density of ~166.3 Wh kg⁻¹. Impressively, this battery system exhibited superior cycling stability over 50 cycles (Fig. 1e).

modified by S, N-codoping and microscopic defects occurrence of

rGO created by the pyridinic N and pyrrolic N synergistically facil-

itate the ions diffusion and provide more active sites for K storage

(Fig. 1b). In addition, the doped S and N can improve the affin-

ity between Sb₂S₃ and graphene. Such rational integration of Sb₂S₃

and S, N-codoped graphene have the following advantages: Firstly,

nanosized active particles shorten the path length for ion transport

over their bulk counterparts. Secondly, graphene layers intimately

contacting with Sb₂S₃ nanoparticles provide express electron path-

way for redox conversion, and meanwhile effectively accommodate

the volume variations and prevent the aggregation of nanoparti-

The above highlighted Sb_2S_3 -SNG composites exhibited remarkable enhancements of both capacity and cycling stability, underlining the rational structure integration of Sb_2S_3 nanoparticles anchored on S, N-codoped graphene, shedding important light on

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https://doi.org/10.1016/j.jechem.2017.12.009

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Fig. 1. (a) HRTEM image of Sb₂S₃-SNG composites; (b) schematic for heteroatomic S and N defects in the SNG; (c) capacity and voltage comparison of Sb₂S₃-SNG with other reported anodes in KIBs; the charge/discharge profles (d) and cycling performance (e) of the Sb₂S₃-SNG/KVPO₄F-C full cell between 0.9–4.6 V at 50 mA g^{-1} [6]. Copyright 2017 Springer.

structure design and interfacial modification for energy storage applications.

Acknowledgments

This work was supported by the National Key Research and Development Program (No. 2016YFA0202500).

Supplementary materials

Supplementary material associated with this article can be found, in the online version, at doi:10.1016/j.jechem.2017.12.009.

Reference

- J.C. Pramudita, D. Sehrawat, D. Goonetilleke, N. Sharma, Adv. Energy Mater. 7 (2017) 1602911.
- [2] X. Zou, P. Xiong, J. Zhao, J. Hu, Z. Liu, Y. Xu, Phys. Chem. Chem. Phys. 19 (2017) 26495–26506.
- [3] Z. Hu, Q. Liu, S.L. Chou, S.X. Dou, Adv. Mater. 29 (2017) 1700606.
- [4] H. Yuan, L. Kong, T. Li, Q. Zhang, Chin. Chem. Lett. 28 (2017) 2180–2194.
 [5] V. Lakshmi, Y. Chen, A.A. Mikhaylov, A.G. Medvedev, I. Sultana, M.M. Rahman, O. Lev, P. Prikhodchenko, A.M. Glushenkov, Chem. Commun. 53 (2017) 8272–8275.
- [6] Y. Lu, J. Chen, Sci. China Chem. 60 (2017) 1533–1539.